Electroless plating of NiP₂ on GO surface for enhanced photocatalytic water splitting

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Abstract

In this investigation, we synthesized a layered graphene oxide @ nickel phosphide (NiP₂@GO) nanocomposites using electroless plating method. Nanocomposite had particle-size of 24nm with the bandgap energy of 2.39 eV and 14 number of layers. The photocatalytic water splitting activity of the nanocomposites were determined in a aqueous hole scavenger electrolyte (20% CH₃OH) solution under a 300W Xe light source irradiation. As-synthesized nanocomposites (NiP₂@GO) exhibited the tenfold increase in photocatalytic H₂ production rate of 408.947 μ molg⁻¹h⁻¹ (apparent quantum efficiency =1.64% at 420nm) over the same for pristine GO sample 40.894 µmolg⁻¹h⁻¹. The intensification photocatalytic activity is attributed to the formation of a p-n junction between rGO and NiP₂ that facilitate the charge transfer and suppressing recombination. The NiP₂ incorporation to GO not only enhances its light harvesting power but also ensure the smooth charge transfer throughout the GO network. Advanced analytical techniques (XRD pattern, UV-Vis spectra, FE-SEM, EDX, steady state PLE spectra, Raman, ESR, XPS, SPV, EIS, etc.) was used to revealed the profound insights into the photocatalytic electron transfer mechanism of the water splitting process wherein the cocatalyst NiP2 accelerates the separation of photo e⁻- h⁺ pairs, enabling efficient relay of photoelectron to reduce H₂O for hydrogen production.

Keywords: Electroless plating, NiP2, graphene oxide, photocatalysis, water splitting, H₂ generation.



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